

A FACILE DEHYDRODIPEPTIDE SYNTHESIS BY THE COUPLING  
BETWEEN TWO  $\alpha$ -DEHYDROAMINO ACIDS

Chung-gi Shin,<sup>\*</sup> Yasuchika Yonezawa, and Juji Yoshimura<sup>†</sup>

Laboratory of Organic Chemistry, Kanagawa University, Kanagawaku, Yokohama 221

<sup>†</sup>Laboratory of Chemistry for Natural Products, Tokyo Institute of Technology,  
Midoriku, Yokohama 227, Japan

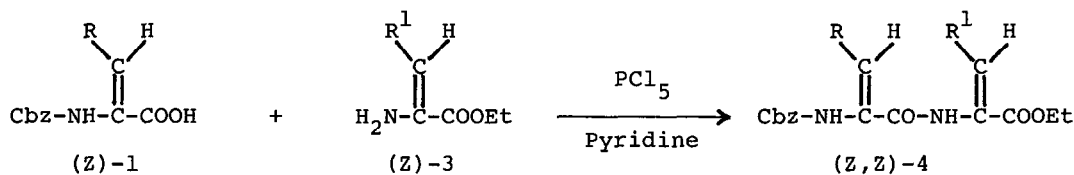
**Summary:** The direct coupling of N-Cbz- $\alpha$ -dehydroamino acid with N-free- $\alpha$ -dehydroamino acid ethyl ester was carried out by the acid chloride method to give a number of dehydrodipeptides.

Recently, increasing interests are being shown in the structure, biological activity, and synthesis of dehydropeptide (DHP), containing one or more  $\alpha$ -dehydroamino acid (DHA) residues, which are very important constituents in a number of antibiotic and phytotoxic peptides.<sup>1-3)</sup>

In the preceding paper,<sup>4)</sup> we reported that the direct coupling of L- $\alpha$ -amino acid ester with N-benzyloxycarbonyl (Cbz)-DHA (1) as a carboxyl component by the mixed anhydride and DCC procedures gave various  $\alpha$ -Cbz-amino-(Z)- $\alpha$ -alkenoylamino acid esters (2). Here, we describe a facile synthesis of DHP by one pot coupling between two different DHA.

To a solution of (Z)-1 (10 mmol) in dry THF (20 ml) was successively added portionwise with stirring  $\text{PCl}_5$  (11 mmol) below 5°C, and then after 20 minutes a chilled solution of ethyl (Z)- $\alpha$ -amino- $\alpha$ -alkenoate (3; 10 mmol)<sup>5)</sup> in dry pyridine (15 ml) also dropwise below 5°C. After the reaction mixture was stirred for 3 h at room temperature, and then poured into ice-water (100 ml). The resulting aqueous solution was extracted three times with ethyl acetate (200 ml). The extracts were successively washed once with 3M-HCl and three times with water, and then dried over anhydrous  $\text{Na}_2\text{SO}_4$ . After evaporation of ethyl acetate under

reduced pressure, the residual syrup was purified on a silica gel column using benzene-ethyl acetate (8 : 1 v/v) as eluent to give a pure syrup or crystals in ca. 50% yield. The compounds obtained were assigned to be ethyl  $\alpha$ -Cbz-amino-(Z)- $\alpha$ -alkenoylamino-(Z)- $\alpha$ -alkenoate (4).



R = alkyl group. Cbz =  $\text{COOCH}_2\text{C}_6\text{H}_5$

Table 1. Yields, physical constants, and spectral data of (Z,Z)-4.

R	R <sup>1</sup>	Yield (%)	Mp °C	IR, cm <sup>-1</sup> C=C	NMR, $\delta$ in CDCl <sub>3</sub>		
					R-CH= (J <sub>Hz</sub> )	R <sup>1</sup> -CH= (J <sub>Hz</sub> )	NH
CH <sub>3</sub>	C <sub>2</sub> H <sub>5</sub>	41	syrup <sup>a</sup>	1680	6.77q (7.5),	6.68t (7.5),	7.30, 8.60
i-C <sub>3</sub> H <sub>7</sub>	C <sub>2</sub> H <sub>5</sub>	43	syrup <sup>a</sup>	1685	6.78d (10.5),	6.42t (7.5),	6.80, 7.82
C <sub>2</sub> H <sub>5</sub>	n-C <sub>3</sub> H <sub>7</sub>	51	syrup <sup>a</sup>	1685	6.70t (7.5),	6.50t (7.5),	6.88, 7.70
CH <sub>3</sub>	i-C <sub>3</sub> H <sub>7</sub>	65	67-68 <sup>b</sup>	1680	6.70q (7.5),	6.60d (10.5),	6.80, 7.70

a) Colorless syrup. b) Colorless needles from cyclohexane. c) Recorded in KBr.

Moreover, another preparative method of 4 has been investigated utilizing the base catalyzed  $\beta$ -elimination of (Z)-2 with a hydroxyl or mercapto group as a leaving group. The results will be reported and discussed elsewhere.

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